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Principles Of Biochemistry Problems

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And Solutions

Biochemistry 3300 Problems (and Solutions) Metabolism I (11) Place an 'x' on each of the high energy bonds shown in the structures below. (12) Consider the oxidation-reduction reaction between cytochrome b and ubiquinol.

Biochemistry 3300 Problems (and

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Solutions) Metabolism I

1.) In lab, you're having fun with buffers based on a 0.1 M solution of sodium phosphate at pH 7.0. In this system, there are three pKAs: 12.7, 7.21, and 2.12. A. (10 points) You may recall that in the pre-lab lecture on the first day, you were shown a chart indicating the relative amounts of different forms of

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phosphate (H_3PO_4 , H_2PO_4^- , HPO_4^{2-} ,

**Chemistry/Biology 302 -
Biochemistry: Exam 1 Practice
Problems**

Solutions for Practice Problems for
Biochemistry, 6: Glycolysis, Respiration
and Fermentation Question 1 ADP ATP
NAD + NADH Glycolysis can be

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summarized as: glucose pyruvate a) In a yeast cell, what is the fate of the carbon in pyruvate under anaerobic conditions? ethanol

Solutions for Practice Problems for Biochemistry, 6

Solutions to Practice Problems for Biochemistry, Session 3:

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Macromolecules: Lipids, Carbohydrates, Nucleic Acid Question 1 The following structure is found in many biological systems.

Solutions to Practice Problems for Biochemistry, Session 3

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In this problem, a 10^{-10} M solution of HCl contributes 10^{-10} M $[H^+]$. The ionization of water contributes 10^{-7} M $[H^+]$. The effective $[H^+]$ of this solution is 10^{-7} M and the $pH=7$. What is the pH of a 10^{-10} M ... BioChemistry

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Proline does not fit well into the structure of the α helix, because of both its geometry (see Problem 29) and the absence of a peptide -NH to contribute

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to hydrogen bonding (see Problem 37).

This amino acid substitution would produce a protein with decreased stability, which would affect the ability of red blood cells to deform their shape in order to squeeze through capillaries.

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