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CHEMISTRY Matter and Change . Section 10.1 Measuring Matter Section 10.2 Mass and the Mole Section 10.3 Moles of Compounds Section 10.4 Empirical and Molecular Formulas Section 10.5

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10 161 Section 10.1 Measuring Matter page 320-324 Practice Problems pages 323-324 1. Zinc (Zn) is used to form a corrosion-inhibiting surface on galvanized steel. Determine the number of Zn atoms in 2.50 mol of Zn. 2.50 mol Zn

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In this chapter you will learn- Concept of matter. Matter in our surroundings (full chapter). Matter in our surroundings Notes. Notes and explanation. Short

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44 . Name Date CHAPTER FOR Class
Section 8.2 continued 9 What is the
relationship between lattice energy and
the strength of the attractive force
holding ions in place? a. The more
positive the lattice energy is, the greater
the force. b. The more negative the
lattice energy is, the greater the force.

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CHAPTER 1 Matter and Change

CCHEMISTRYHEMISTRY MMATTER AND CHANGEATTER AND CHANGE Consultant

Douglas Fisher, PhD

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