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increase in the number of moles of gas;
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$2.4 \times 10^{-9} > 1.4 \times 10^{-16}$ AND precipitate
will form If value given is used:

ALTERNATIVE 3: $Q [Pb^{2+}] [OH^{-}]^2 = 2.4 \times 10^{-4} (1.0 \times 10^{-2})^2 = 2.4 \times 10^{-8} > K_{sp}$
AND precipitate will form OR $2.4 \times 10^{-8} > 1.43 \times 10^{-20}$ AND precipitate will form

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