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their atomic numbers so that elements with similar properties fall in the same column, or group.

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Review. Describe.... 1)

Dalton's model - solid
sphere, made up of
only 1 type of matter,
no space. 2)

Schrodinger's model -
based on probability of
finding an e-; electrons
are found in orbitals

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(look like electron clouds), which are in sublevels (s,p,d,f,g), which are in energy levels.

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elements with the atomic numbers 58-71. The 14 elements with the atomic numbers 90-103. The elements of group 1 of the periodic table.

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CHAPTER 5 REVIEW

The Periodic Law

SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1.

c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were

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discovered.

5 The Periodic Law

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